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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/563,815	01/06/2006	Tetsuro Mizushima	283358US0PCT	1896

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OBLON, SPIVAK, MCCLELLAND MAIER & NEUSTADT, P.C.
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EXAMINER

ANGEBRANDT, MARTIN J

ART UNIT	PAPER NUMBER
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1795

NOTIFICATION DATE	DELIVERY MODE
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12/19/2008

ELECTRONIC

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

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Office Action Summary	Application No. 10/563,815	Applicant(s) MIZUSHIMA, TETSURO	
	Examiner Martin J. Angebrannt	Art Unit 1795	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 6/13/08 or 9/11/08.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-16 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1-16 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
 2. ☐ Certified copies of the priority documents have been received in Application No. _____.
 3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- | | |
|----------------------------------------------------------------------------------------|-------------------------------------------------------------------|
| 1) <input checked="" type="checkbox"/> Notice of References Cited (PTO-892) | 4) <input type="checkbox"/> Interview Summary (PTO-413) |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948) | Paper No(s)/Mail Date. _____ |
| 3) <input checked="" type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08) | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| Paper No(s)/Mail Date <u>6/13/08</u> . | 6) <input type="checkbox"/> Other: _____ |

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1. Those references cited by the applicant, but marked as not considered have been cited by the examiner making their PGpub of record. The response of the applicant has been read and given careful consideration. Responses to the arguments are presented after the first rejection to which they are directed

2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

3. Claims 1-10,12-14 and 16 are rejected under 35 U.S.C. 103(a) as being unpatentable over Chandross et al. EP 938027, in view of Loy, et al., "Substituent effect on the sol-gel chemistry of organotrialkoxysilanes", Chem. Mater., Vol. 12(12) pp. 3624-3632 (2000), Shustack et al. '566 and Hiruma et al., '831.

Chandross et al. EP 938027 teaches in example 1, the matrix precursor comprising methyltriethoxysilane and phenyltriethoxysilane combined with 10 wt % lauryl acrylate and photoinitiator CGI-784 and hydrolyzed to form the a holographic recording medium and coated to the thickness of 300 microns. This was used in example 3 to record a holograms [0028-0034]. The matrix precursor is $R_n(M)(OR')_{4-n}$ where R is aryl or alkyl, R' is lower alkyl, n is 1 or 2 and M is Si, Ti, Ge, Zr, V or Al and is hydrolyzed to form the form the matrix. [0009,0014-0018,0022-0025]. Organic moieties are disclosed as increasing compatibility with the photopolymer and the free volume/porosity. [0011]. The use of various monomers including isobornyl acrylate, phenoxyethyl acrylate [0019]. The use of organoalkoxysilanes, and particularly trialkoxysilanes is disclosed [0022].

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Loy, et al., "Substituent effect on the sol-gel chemistry of organotrialkoxysilanes", Chem. Mater., Vol. 12(12) pp. 3624-3632 (2000) teaches the formation of organic/inorganic materials/xerogels/matrices. $R(M)(OR')_3$, where R' is methyl or ethyl and R is hydrogen, methyl, ethyl, propyl, butyl, isobutyl, t-butyl, hexyl, octyl, decyl, dodecyl, hexadecyl, octyl, octadecyl, cyclohexyl, vinyl, phenyl, benzyl, phenethyl, chloromethyl, p-chloromethylphenyl or tridecafluoro-1,1,2,2-tetrahydrooctyl (page 3624). The chloromethyl and chloromethylphenyl are shown to react more slowly than the methyl in table 5. (page 3628). The pore diameters for the chloromethyl is significantly larger than that for the methyl substituted silane as shown in tables 7 and 8 (page 3631). The hydrogen, methyl and chloromethyl gels were the most transparent (page 3631, right column). The formation of gels using materials usually used as coupling agents is disclosed (page 3632, left column)

Shustack et al. '566 teaches organometallic coupling agents such as (bromophenyl)trimethoxysilane and (chlorophenyl)trimethoxysilane lowers scattering losses. [0025].

Hiruma et al., '831 teaches coupling agents increase the adhesion between photocurable coatings and the underlying substrate. Useful coupling agents include chloropropyl trimethoxysilane (3/61-4/66).

It would have been obvious to one skilled in the art to modify the examples of Chandross et al. EP 938027 by replacing the methyltriethoxysilane with chloromethyltriethoxysilane as taught by Loy, et al., "Substituent effect on the sol-gel chemistry of organotrialkoxysilanes", Chem. Mater., Vol. 12(12) pp. 3624-3632 (2000) with a reasonable expectation of forming useful organic-inorganic matrix holographic recording medium, based upon the transparency

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disclosed by Loy, et al., "Substituent effect on the sol-gel chemistry of organotrialkoxysilanes", Chem. Mater., Vol. 12(12) pp. 3624-3632 (2000) with increase compatibility with the organic matrix based upon the teaching of the coupling functionality of chlorinated alkyl/aryl trialkoxysilanes by Shustack et al. '566 and Hiruma et al., '831.

The showing of the instant specification is limited to the use of mixtures of TEOS and the inventive composition in a particular ratio, acid cured and with the high refractive index monomers used. The claims are quite a bit broader than this. Further it is known from Loy, et al., "Substituent effect on the sol-gel chemistry of organotrialkoxysilanes", Chem. Mater., Vol. 12(12) pp. 3624-3632 (2000) that The hydrogen, methyl and chloromethyl gels were the most transparent and from Chandross et al. EP 938027 that the presence of the $R_n(M)(OR')_{4-n}$ where R is alkyl or aryl yield matrices which are more organic compatible. The examples of the instant specification in using a mixture of TEOS and the inventive silane has clearly reduced the organic character of the matrix and therefore its compatibility with the photopolymer.

The applicant is correct that no one of the references teach the invention claimed, but this position fails to account for the fact that the rejection relies upon several references. As the matrix which forms a substantial portion of the composition is more transparent, as discussed in Loy et al. it is reasonable to expect that the modification of using the chloromethyl or chloromethylphenyl containing would confer at least some increased transparency to the holographic article which would be desirable. Further the reduced scattering discussed by Shustack is clearly a benefit which relates to the performance of the holographic media as noted on page 7 of the response in the first paragraph. As the light passes through the hologram and it is an optical article, transparency is clearly a desirable attribute and no specific direction in

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Chendross et al. need be present for one of ordinary skill in the art to appreciate this, but Chandross does discuss that the cured samples in example 1 are “transparent” and indicates that this is a desirable property. The applicant argues that there is no motivation to reduce scatter. The discussion of low levels of bubbles and other internal imperfections in examples 1 addresses this as these would clearly result in scattering of light and one of ordinary skill in the art would recognize that light which is scattered is not able to be diffracted to form the holographic image. The issue of adhesion to an underlying substrate is also clear for one of ordinary skill in the art, noting that the holographic recording composition is provided between two substrates in Chandross et al. The applicant argues that they have shown an increased compatibility between the cured matrix of the invention and the photocurable components. The examiner notes that only one photosensitive composition has been used which is a mixture of phenoxyethyl acrylate, 2,2-bis(4-acryloxy-diethoxy)phenyl]propane and irgacure 784. This is not a showing of sufficient breadth for the coverage sought, noting all the free radically polymerizable monomers listed at [0043] of the prepub and the cationically curable monomers discussed at [0048-0051]. The showing of only for a combination of two free radically polymerizable monomers. For the scope of coverage sought, more data would have to be of record. With respect to separation, perhaps the issue of adhesion discussed by Hiruma et al. plays a part. The rejection stands.

4. Claims 1-10 and 12-16 are rejected under 35 U.S.C. 103(a) as being unpatentable over Chandross et al. EP 938027, in view of Loy, et al., “Substituent effect on the sol-gel chemistry of organotrialkoxysilanes”, Chem. Mater., Vol. 12(12) pp. 3624-3632 (2000), Shustack et al. ‘566 and Hiruma et al., ‘831 further in view of Otaki et al. ‘740.

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Otaki et al. teaches the holographic recording medium with an inorganic-organic matrix. The photoreactive monomer may be used in amounts of 10-1000 part to 100 parts of the organic-inorganic polymer, preferably 10-100 parts per 100 parts. [0090]

To address other embodiments bounded by the claims, but not rendered obvious above, the examiner cites Otaki et al. '740 and holds that it would have been obvious to modify the media rendered obvious by the combination of Chandross et al. EP 938027, Loy, et al., "Substituent effect on the sol-gel chemistry of organotrialkoxysilanes", Chem. Mater., Vol. 12(12) pp. 3624-3632 (2000), Shustack et al. '566 and Hiruma et al., '831 by using a higher relative amount of monomer, such as 50-1000 parts per 100 parts of matrix with a reasonable expectation of successfully forming a useful holographic recording medium based upon the teachings of Otaki et al. '740.

5. Claims 1-16 are rejected under 35 U.S.C. 103(a) as being unpatentable over Chandross et al. EP 938027, in view of Loy, et al., "Substituent effect on the sol-gel chemistry of organotrialkoxysilanes", Chem. Mater., Vol. 12(12) pp. 3624-3632 (2000), Shustack et al. '566, Hiruma et al., '831 and Otaki et al. '740, further in view of Poutasse et al. '782 or Zeldin et al. '172.

Poutasse et al. '782 teaches silanes which act as adhesion promoters (silane coupling agents) bounded by formula B including 8-bromooctyl trimethoxy silane, bromophenyl trimethoxy silane, 3-bromopropyl trimethoxysilane, 2-chloroethyl triethoxy silane, p-chloromethylphenyl trimethoxy silane, chloromethyl triethoxy silane, 3-chloropropyl triethoxy silane and iodopropyl trimethoxy silane (6/32-7/57).

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Zeldin et al. '172 teaches various silane coupling agents including bromoethyltriethoxy silane, chloroethyltriethoxy silane, bromopropyltriethoxy silane, bromopropyltrimethoxy silane, bromomethyl triethoxysilane, and the like (3/2-57).

In addition to the basis above, it would have been obvious to modify the media rendered obvious by the combination of Chandross et al. EP 938027, Loy, et al., "Substituent effect on the sol-gel chemistry of organotrialkoxysilanes", Chem. Mater., Vol. 12(12) pp. 3624-3632 (2000), Shustack et al. '566, Hiruma et al., '831 and Otaki et al. '740 by using other silane coupling agents, such as those bounded by formula B and/or taught by Poutasse et al. '782 or those used by Zeldin et al. '172, in place of the chloromethyl trialkoxides exemplified by Loy, et al., "Substituent effect on the sol-gel chemistry of organotrialkoxysilanes", Chem. Mater., Vol. 12(12) pp. 3624-3632 (2000), Shustack et al. '566 and Hiruma et al., '831 with a reasonable expectation of gaining the benefits disclosed in adhesion and the like based upon these being silane coupling agents/adhesion promoters.

6. **THIS ACTION IS MADE FINAL.** Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event,

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however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

7. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Martin J. Angebranntdt whose telephone number is 571-272-1378.

The examiner can normally be reached on Monday-Friday.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mark Huff can be reached on 571-272-1385. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/Martin J Angebranntdt/
Primary Examiner, Art Unit 1795

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Primary Examiner
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12/12/2008